Two Halogenated Sesquiterpenoids from the Fruits of Alpinia oxyphylla

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Two new halogenated eremophilane-type sesquiterpenoids, (11S)- and (11R)-12-chloronootkaton-11-ol (1 and 2, resp.), together with five known sesquiterpenoids, nootkatone (3), 7-epiteucrenone B (4), oxyphyllenodiol A (5), oxyphyllenodiol B (6), and alpinenone (7), were isolated from the EtOH extract of the fruits of *Alpinia oxyphylla* MIQ. The structures were elucidated based on the analyses of their spectroscopic data.

Introduction. – The fruits of *Alpinia oxyphylla* MIQ. (Zingiberaceae), '*Yizhiren*' in Chinese, has been used as a traditional Chinese Medicine for the treatment of diarrhea, diuresis, and dementia [1]. In recent years, various types of sesquiterpenoids have been isolated from this plant, and some of them showed inhibitory effects on nitric oxide (NO) production in lipopolysaccharide (LPS)-activated mouse peritoneal macrophages or IFN- γ -induced RAW 264.7 murine macrophages [2–6]. In our present study, two new halogenated eremophilane-type sesquiterpenoids, (11*S*)- and (11*R*)-12-chloronootkaton-11-ol (**1** and **2**, resp.), together with five known sesquiterpenoids, nootkatone (**3**) [7], 7-epiteucrenone B (**4**) [8], oxyphyllenodiol A (**5**) [2], oxyphyllenodiol B (**6**) [2], and alpinenone (**7**) [9] (*Fig. 1*), were isolated from the EtOH extract of the fruits of *A. oxyphylla*. Compound **7** was isolated from *A. oxyphylla* for the first time. Herein, we report the isolation and structure elucidation of the new compounds **1** and **2**.



Fig. 1. Structures of compounds 1-7 isolated from A. oxyphylla

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Results and Discussion. - Compounds 1 and 2 were separated by prepative HPLC and obtained as colorless oils. They showed yellow or orange color under sunlight, and bright-yellow fluorescence under UV 365 nm when sprayed with 10% H₂SO₄ or 5% vanillin/ H_2SO_4 . The ESI mass spectra of compounds 1 and 2 exhibited the same characteristic *pseudo*-molecular-ion clusters at 271.3/273.4 $(3:1, [M + H]^+)$ and 293.3/ 295.4 (3:1, $[M + Na]^+$), indicating that both of them contain one Cl-atom. The molecular formula of 1 was determined as $C_{15}H_{23}ClO_2$, on the basis of HR-ESI-MS signals at m/z 271.1464 ($[M + H]^+$) and 293.1281 ($[M + Na]^+$), indicating four degrees of unsaturation. Compound 2 had the same molecular formula according to its HR-ESI-MS signals at m/z 271.1463 ($[M + H]^+$) and 293.1279 ($[M + Na]^+$). The ¹H-NMR spectrum (*Table 1*) of **1** displayed signals of three Me groups at $\delta(H)$ 1.00 (d, J = 4.2), and at $\delta(H)$ 1.14 (s) and 1.21 (s), of one Cl-bearing CH₂ group at $\delta(H)$ 3.65 and 3.70 (2d, J = 10.8, both 1 H), of one OH group at $\delta(H)$ 3.79 (s), and of one olefinic H-atom at $\delta(H)$ 5.67 (s). The ¹H-NMR signals of **2** were similar to those of **1**, with slight differences in some chemical shifts (Table 1). The ¹³C-NMR spectrum (Table 2) of 1 displayed signals for 15 C-atoms, including those of one CO C-atom at $\delta(C)$ 197.6, two olefinic C-atoms at $\delta(C)$ 169.8 and 124.0, one O-bearing quaternary C-atom at $\delta(C)$ 72.7, one Cl-bearing CH₂ group at δ (C) 53.0, and three Me groups at δ (C) 14.3, 16.3, and 21.3. As shown in *Table 2*, except for the little differences in chemical shifts of C(5), C(6), C(7), and C(8), the remaining ¹³C-NMR signals of **2** were almost coincident with those of 1.

Table 1. ¹*H-NMR Spectral Data of* **1** and **2**. At 600 MHz, in (CD₃)₂CO; δ in ppm, *J* in Hz. Atom numbering as indicated in *Fig. 1*.

H-Atom	1	2	
H–C(1)	5.67 (br. <i>s</i>)	5.67 (br. s)	
$H_{a-}C(3)$	2.07 - 2.10 (m)	2.08 - 2.10 (m)	
$H_{\beta}-C(3)$	2.29 (t, J = 15.6, 15.6)	2.29 (t, J = 15.0, 16.2)	
$H_{\beta}-C(4)$	1.94 - 2.01 (m)	1.97 - 2.03 (m)	
$H_{a-}C(6)$	2.01 - 2.05 (m)	2.19 (br. $d, J = 13.2$)	
$H_{\beta}-C(6)$	1.08 (t, J = 13.8, 12.6)	1.12 - 1.17 (m)	
H-C(7)	2.10-2.15(m)	2.10-2.14(m)	
$H_a - C(8)$	2.09-2.13(m)	1.94 (br. $d, J = 18.6$)	
$H_{\beta}-C(8)$	1.29 (dd, J = 12.6, 12.6)	1.25 - 1.30 (m)	
$H_{a-}C(9)$	2.54 (t, J = 15.0, 13.8)	2.56 (t, J = 14.4, 14.4)	
$H_{\beta}-C(9)$	2.41 (br. $d, J = 15.0$)	2.39 (br. $d, J = 15.0$)	
$CH_{2}(12)$	3.65 (d, J = 10.8), 3.70 (d, J = 10.8)	3.63 (d, J = 10.8), 3.68 (d, J = 10.8)	
Me(13)	1.21 (s)	1.21(s)	
Me(14)	1.00 (d, J = 4.2)	0.99 (d, J = 6.0)	
Me(15)	1.14 (s)	1.13 (s)	
HO-C(11)	3.79 <i>(s)</i>	3.82 (s)	

Correlations in both of the ¹H,¹H-COSY spectra of **1** and **2** indicated the presence of two spin systems, including $-CH_2-CH-Me$ and $-CH_2-CH_2-CH_2-CH_2-$ fragments (*Fig.* 2). Both of the HMBC spectra of **1** and **2** exhibited the correlations from CH₂(3) to C(1), C(2), C(4), C(5), and C(14), from Me(15) to C(10), from CH₂(6) to C(4), C(5), Me(15), C(10), C(7), C(8), and C(11), from Me(13) to C(7), C(11), and C(12),

C-Atom	\mathbf{A}^{a})	B ^a)	1 ^a)	1 ^b)	2 ^b)
H-C(1)	124.4 (d)	124.5(d)	124.6 (<i>d</i>)	124.0(d)	123.9 (d)
C(2)	199.9 (s)	199.8 (s)	199.6 (s)	197.6 (s)	197.6 (s)
$CH_{2}(3)$	42.0(t)	42.0(t)	42.0(t)	41.8(t)	41.8(t)
H-C(4)	40.5(d)	40.6(d)	40.5(d)	40.7(d)	40.7 (d)
C(5)	39.6 (s)	39.2 (s)	39.6 (s)	39.5 (s)	39.0 (s)
$CH_{2}(6)$	39.2(t)	38.6(t)	39.2(t)	39.2(t)	38.6(t)
H-C(7)	39.7 (d)	39.2(d)	39.9(d)	39.5(d)	39.2 (d)
$CH_2(8)$	26.8(t)	27.9(t)	26.6(t)	26.6(t)	27.4(t)
$CH_2(9)$	33.0(t)	32.9(t)	32.8(t)	32.6(t)	32.4(t)
C(10)	171.0(s)	170.8(s)	170.1(s)	169.8(s)	169.8 (s)
C(11)	74.3(s)	74.2(s)	73.4(s)	72.7(s)	72.6(s)
$CH_{2}(12)$	68.2(t)	68.4(t)	53.5(t)	53.0(t)	52.8(t)
Me(13)	20.3(q)	19.9(q)	21.2(q)	21.3(q)	21.4(q)
Me(14)	14.9(q)	15.0(q)	14.9(q)	14.3(q)	14.3(q)
Me(15)	16.9(q)	16.8(q)	16.8(q)	16.3(q)	16.1(q)

Table 2. ¹³C-NMR Spectral Data of (11S)- and (11R)-Nootkatone-11,12-diol (8 and 9, resp.) [7], and 1 and 2. At 150 MHz; δ in ppm. Atom numbering as indicated in Fig. 1.



Fig. 2. ¹H,¹H-COSY correlations (—) and key HMBCs ($H \rightarrow C$) of 1 and 2

from CH(7) to C(13), C(11), and C(12), and from $CH_2(9)$ to C(1), C(10), C(5), C(8), and C(7). These findings evidenced that **1** and **2** are eremophilane-type sesquiterpenoids.

The relative configurations of 1 and 2 (Fig. 1) were determined by NOESY experiments. Compound 1 showed NOE correlations H_{β} -C(4)/ H_{β} -C(6), H_{β} -C(3); $H_{\beta}-C(9)/H_{\beta}-C(3), H_{\beta}-C(8); H_{\alpha}-C(6)/Me(14), Me(15); Me(15)/H_{\alpha}-C(7), H_{\alpha}-C(8),$ $H_a-C(6)$; and $Me(13)/H_a-C(7)$, $H_a-C(8)$, and 2 exhibited NOE correlations $H_a-C(9)/Me(14), Me(15), H_a-C(7), H_a-C(8); H_a-C(6)/Me(14), Me(15); H_{\beta}-C(9)/Me(14), Me(15), Me(15),$ H–C(1), H_g–C(6), H_g–C(8); and H_g–C(4)/H_g–C(3), indicating that 1 and 2 were halogenated derivatives of nootkatone (3). The absolute configuration at C(11) was determined through comparison with the structures of the reported eremophilane-type sesquiterpenoids, (11S)- and (11R)-nootkatone-11,12-diol (8 and 9, resp.), which were produced from biotransformation of (+)-nootkatone by Aspergillus niger [7]. As shown in *Table 2*, the C-atom signals of **1** and **2** were very similar to those of **8** and **9**, except for the signal of C(12), which was substituted by a Cl-atom in 1 and 2. As reported, the absolute configurations of 8 and 9 could not be characterized by CD spectra, because both thiocarbonates of 8 and 9 showed negative Cotton effects at ca. 310 nm in CD spectra. The (S)-configuration at C(11) of 8 was finally established based on X-ray crystallographic analysis of the thiocarbonate of $\mathbf{8}$ (Fig. 3) [7]. Interestingly,



Fig. 3. ORTEP Drawing of the thiocarbonate of 8 (from [7])

some regularities can be deduced through comparison of the C-atom signals of **8** and **9**, which may be helpful for the determination of the relative configuration at C(11): the signals of C(5), C(6), and C(7) of **8** appeared at a lower field (+0.4, +0.6, +0.5 ppm, resp.), and the signal of C(8) appeared at a higher field (-1.1 ppm), compared to those of **9** (*Table 2*). Similarly, the signals at C(5), C(6), and C(7) of **1** appeared at a lower field (+0.5, +0.6, +0.3 ppm, resp.), and the signal of C(8) appeared at a lower field (-0.8 ppm) compared to those of **2**, demonstrating that **1** and **2** were (11*S*)- and (11*R*)-12-chloronootkatone-11-ol, respectively.

Experimental Part

General. Column chromatography (CC): silica gel (SiO₂; 200–300 mesh; Qingdao Haiyang Chemical Group Co. Ltd., P. R. China), ODS-C18 (75 µm, YMC Co., Japan), MCI gel (CHP-20P, 75– 150 µ, Mitsubishi Chemical Co., Japan), and Sephadex LH-20 (Pharmacia Fine Chemicals, USA). Semiprep. HPLC: Shimadzu Prominence LC-20A liquid chromatography, with LC-20AT pumps, SPD-20A UV detector (Shimadzu Co., Japan), and a YMC-Pack ODS-A column (250 mm × 10 mm, 5 µm) (YMC Co., Japan). TLC: silica-gel GF254 plates (Qingdao Haiyang Chemical Group Co. Ltd., P. R. China); visualization by heating the plates sprayed with 10% H₂SO₄ or 5% vanillin/H₂SO₄ reagent. UV Spectra: UV-2450 spectrophotometer (Shimadzu Co., Japan). Optical rotations: Gyromat-Hp digital automatic polarimeter (Kernchen Co., Germany). CD Spectra: Chirascan spectrophotometer (Applied Photophysics Ltd., UK). NMR Spectra: Bruker AV-600, δ in ppm, with TMS as an internal standard, J in Hz. ESI-MS Spectra: API 4000 mass spectrometer (Applied Biosystems Co., USA). HR-ESI-MS: LTQ-Orbitrap mass spectrometer (Thermo Fisher Co., USA), in m/z.

Plant Material. The fruits of Alpinia oxyphylla MIQ. were purchased from Shuyupingmin Pharmacy (Jinan, P. R. China) in March 2011 and authenticated by L. X., School of Pharmacy, Shandong

University, P. R. China. A voucher specimen (No. 090801) has been deposited with the Laboratory of Pharmacognosy, School of Pharmacy, Shandong University.

Extraction and Isolation. The dried fruits of A. oxyphylla (5 kg) were powdered and extracted with 95% EtOH (4 \times 25 l) at r.t. The combined extracts were evaporated to yield a crude extract (450 g), which was then suspended in 20% EtOH and partitioned with petroleum ether (PE), AcOEt, and BuOH successively. The AcOEt extract (210 g) was separated by CC (SiO₂, PE/AcOEt 20:1 to 1:1) to yield twelve fractions, Frs. 1-12. Fr. 3 (30.5 g) was subjected to CC (Sephadex LH-20; EtOH) to yield three fractions Frs. 3.1-3.3. Fr. 3.2 (5.2 g) was purified by CC (1. SiO₂, PE/CH₂Cl₂ 1:9; 2. ODS-C18, EtOH/ H_2O 1:1 to 4:1) to afford compound 3 (700 mg). Fr. 5 (1.6 g) was subjected to CC (MCI, EtOH/ H_2O 35:65 to 70:30) to afford eight subfractions, Frs. 5.1-5.8. Compound 7 (5 mg) was obtained from recrystallisation of Fr. 5.3. Fr. 8 (11.1 g) was subjected to CC (ODS-C18, EtOH/H₂O 2:8 to 7:3) to yield nine subfractions, Frs. 8.1-8.9. Fr. 8.2 (1.2 g) was further purified by CC (1. MCI, EtOH/H₂O 24:76 to 7:3; 2. SiO₂, CH₂Cl₂/AcOEt 20:1 to 4:1) to yield 4 (28.1 mg). Fr. 8.3 (750 mg) was purified by CC (SiO₂, $CH_2Cl_2/AcOEt\ 20:1\ to\ 5:1)\ and\ semi-prep.\ HPLC\ (MeOH/H_2O\ 6:4)\ to\ yield\ 1\ (9.5\ mg)\ and\ 2\ (4.7\ mg).$ Fr. 9 (25 g) was separated by CC (SiO₂, CH₂Cl₂/AcOEt 100:1 to 1:1) to afford ten subfractions, Frs. 9.1 – 9.10. Fr. 9.4 (5.1 g) was purified by repeated CC (SiO₂, CH₂Cl₂/AcOEt) to furnish twelve subfractions, Frs. 9.4.1–9.4.12. Compound 5 (72.6 mg) was obtained from recrystallization of Fr. 9.4.1. Compound 6 (79.3 mg) was obtained from Fr. 9.4.2 by semi-prep. HPLC (MeOH/H₂O 75:25).

(11S)-12-Chloronootkaton-11-ol (=(4R,4aS,6R)-6-[(2S)-1-Chloro-2-hydroxypropan-2-yl]-4,4a,5,6,7,8-hexahydro-4,4a-dimethylnaphthalen-2(3H)-one; **1**). Colorless oil. $[\alpha]_{D}^{20} = +41.0$ (c = 0.1, MeOH). UV (MeOH): 238 (4.72). CD (c = 0.1, MeOH): 317 (-0.66). ¹H- and ¹³C-NMR: see *Tables 1* and 2, resp. HR-ESI-MS: 271.1464 ($[M + H]^+$, C₁₅H₂₄³⁵ClO²₂; calc. 271.1465), 293.1281 ($[M + Na]^+$, C₁₅H₂₃³⁵ClNaO²₂; calc. 293.1284).

(11R)-12-Chloronootkaton-11-ol (=(4R,4aS,6R)-6-[(2R)-1-Chloro-2-hydroxypropan-2-yl]-4,4a,5,6,7,8-hexahydro-4,4a-dimethylnaphthalen-2(3H)-one; **2**). Colorless oil. $[a]_{D}^{2D} = +6.0$ (c = 0.1, MeOH). UV (MeOH): 238 (3.93). CD (c = 0.1, MeOH): 312 (-0.19). ¹H- and ¹³C-NMR: see *Tables 1* and 2, resp. HR-ESI-MS: 271.1463 ($[M + H]^+$, C₁₅H₂₄³⁵ClO₂⁺; calc. 271.1465), 293.1279 ($[M + Na]^+$, C₁₅H₂₃³⁵ClNaO₂⁺; calc. 293.1284).

The crystal data for the thiocarbonate of (11*S*)-nootkatone-11,12-diol (**8**; from [7]): monoclinic, space group P_{2_1} , a = 8.710 (0) Å, b = 8.448 (0) Å, c = 10.988 (0) Å, $\beta = 102.428$, V = 789.599976 (0) Å³, Z = 2, $D_x = 1.530$ Mg/m³, $D_m = 1.500$ Mg/m³, μ (Mo K_a) = 2.09 mm⁻¹, Eta; +1.9. Final *R* and R_w were 0.044 and 0.074, resp., for 1186 reflections.

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